mullite appears in the charge. Transformation of cristobalite into quartz reaches maximum intensity at the highest pressure (2000 kG/cm²).

<u>Palygorskite</u>. In 22 to 48 hour runs decomposition of palygorskite is completed sooner at high pressures, although, as in the case of sepiolite, the process begins at 325° C^{*} Beginning with a temperature of 325° C, in the entire temperature range, palygorskite was transformed into montmorillonite, whose appearance is indicated on the diffractometer traces by reflection $\underline{d}_{001} = 14.7$ A (Fig.2<u>B</u>, e,f,d). In samples saturated with glycerine \underline{d}_{001} increased to 38 A (Fig.3<u>A</u>, a,b,c), and after annealing at 600° C, decreased to 9.9 A (Fig.3<u>B</u>,a,b,c).

The montmorillonite formed from palygorskite is dioctahedral with $\underline{d}_{060} = 1.485$ A (Fig.2<u>B</u>, d). The trnasformation may be represented by equation:

(Mg1.78E00.35A11.22)(Si7.41A10.59)018.96(OH)5.28(OH2)1.76 --

-- $(Mg_{1.78}Fe_{0.35}Al_{1.22})(Si_{7.41}Al_{0.59})O_{20}(CH)_4 + 2H_2O$ Beginning at 500°C, under water vapor pressure, montmorillonite

* In the experiments at 100-200°C (Fig. 2B, b',c') palygorskite with admixed calcite was used, but in the experiments at higher temperatures (Fig.2B, d-1) only palygorskite freed of calcite by washing in 5% HCl (Fig.2B,a) was used.

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